# Characterization of active-site aromatic residues in xylanase A from *Streptomyces lividans*

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The role of four aromatic residues (W85, Y172, W266 and W274) in the structure-function relationship in xylanase A from *Streptomyces lividans* (XlnA) was investigated by site-directed mutagenesis where each residue was subjected to three substitutions (W85A/H/F; W266A/H/F; W274A/ H/F and Y172A/F/S). These four amino acids are highly conserved among family 10 xylanases and structural data have implicated them in substrate binding at the active site. Far-UV circular dichroism spectroscopy was used to show that the overall structure of XlnA was not affected by any of these mutations. High-performance liquid chromatographic analysis of the hydrolysis products of birchwood xylan and xylopentaose showed that mutation of these aromatic residues did not alter the enzyme's mode of action. As expected, though, it did reduce the affinity of XInA for birchwood xylan. A comparison of the kinetic parameters of different mutants at the same position demonstrated the importance of the aromatic nature of W85, Y172 and W274 in substrate binding. Replacement of these residues by a phenylalanine resulted in mutant proteins with a  $K_{\rm M}$  closer to that of the wild-type protein in comparison with the other mutations analyzed. The kinetic analysis of the mutant proteins at position W266 indicated that this amino acid is important for both substrate binding and efficient catalysis by XlnA. These studies also demonstrated the crucial role of these active site aromatic residues for the thermal stability of XlnA.

*Keywords*: family 10 hydrolase/glycosyl hydrolase/sitedirected mutagenesis/structure-function/xylanase

#### Introduction

Xylanases (EC 3.2.1.8) catalyze the hydrolysis of xylan, a complex heteropolymer found in plant cell walls. On the basis of sequence homology of their catalytic domains, xylanases have been classified into families 10 and 11 of the 66 families of glycosyl hydrolases (Henrissat and Bairoch, 1998). Xylanase A from *Streptomyces lividans* (XlnA) is a 47 kDa secreted protein composed of two domains: a 33 kDa catalytic domain responsible for the xylanolytic activity and a 14 kDa xylan-binding domain involved in the specific binding to the insoluble substrate (Dupont *et al.*, 1998). The catalytic domain of XlnA belongs to family 10 of glycosyl hydrolases (Shareck *et al.*, 1991) and folds into an  $(\alpha/\beta)_8$  motif (Derewenda *et al.*, 1994). The catalytic domain of XlnA, along with members of 11

other different families of glycosyl hydrolases, is also classified into clan GH-A on the basis of structural and mechanistic information (Henrissat and Bairoch, 1998). Xylanases hydrolyze  $\beta$ -1,4-glycosidic bonds with net retention of the configuration of the anomeric carbon via a double displacement mechanism (Sinnott, 1990). The side chains of two glutamic acids are involved, one acting as a nucleophile and the other as an acid–base catalyst. These two amino acids were identified in XlnA by site-directed mutagenesis (Moreau *et al.*, 1994) and confirmed with the structure of the catalytic domain (Derewenda *et al.*, 1994).

Available structural information on family 10 xylanases has allowed the identification of the residues present in the active site along with their possible involvement in the catalytic function of this type of protein. For example, substrate binding in the active site of family 10 xylanases has been investigated with the use of co-crystal structures. In one of these studies, the structure of XynA from Pseudomonas fluorescens subsp. cellulosa has been obtained in the presence of xylopentaose (Harris et al., 1994). This permitted the identification of the residues forming six substrate-binding subsites (-2 to +4;Davies, et al. 1997) of this enzyme. In another study, White et al. (1996) obtained the crystal structure of Cex from Cellulomonas fimi covalently complexed with a 2-fluorocellobioside, an analogue of the catalytic intermediate in the double displacement mechanism. They identified many interactions which could be of crucial importance in the binding and stabilization of the catalytic intermediate in the active site of family 10 xylanases. All these structural data show the implication of different aromatic residues in the binding of the substrate and/or the catalytic intermediate. The aromatic amino acids would allow specific and stable binding of the carbohydrate moieties by partial stacking of the sugar ring (Quiocho, 1986) and/or by hydrogen bonding to hydroxyl groups of the pyranose ring.

The role of aromatic residues in the function of xylanases has been studied through chemical modification. For example, the inactivation of a family 11 xylanase from *Streptomyces*  $T_7$  with *N*-bromosuccinimide suggested the importance of a tryptophan side chain in the active site of the enzyme (Keskar *et al.*, 1989). Moreover, for family 11 xylanase XynA from *Schizophilum commune*, chemical modification of tyrosyl side chains by tetranitromethane suggested that Y97, a conserved residue in this family of enzyme, plays an essential role in substrate binding (Bray and Clarke, 1995).

The active site of XlnA contains many aromatic residues, four of which are exposed to solvent and could be involved in substrate binding (W85, Y172, W266 and W274). These residues are highly conserved in family 10 glycosyl hydrolases, suggesting their importance in the function of this type of enzyme (Figure 1). Moreover, the equivalents of these amino acids were shown to be part of the substrate-binding subsites surrounding the cleavage site that contain the two catalytic residues, -1 and +1 (Harris *et al.*, 1994). Site-directed muta-

	W85	Y172	W266	W274
	$\downarrow$	$\downarrow$	$\downarrow$	$\downarrow$
A	QVRGHTLA <b>W</b> HS	 AKLCYNDYNVENW	 GITVWGVR	DSDSWRS-
В	LIRGHTLVWHS	 AKLYIND <b>Y</b> NLDSA	 GITV <b>W</b> GVA	DPDS <b>W</b> RA-
С	LIRGHTLVWHS	 AKLYIND <b>Y</b> NLDSA	 GITVWGVA	DPDS <b>W</b> RS-
D	LIRGHTLLWHS	 AKLYIND <b>Y</b> NLDKS	 GITVWGVS	DKDS <b>W</b> RP-
E	QVRGHTLA <b>W</b> HS	 AKLCYNDYNIENW	 GITVWGVR	DTDS <b>W</b> RS-
F	EVRGHTLAWHS	 AKLCYND <b>Y</b> NIENW	 GITVWGVR	DTDSWRS-
G	TVHGHALV <b>W</b> HS	 VILYYND <b>Y</b> NIE-Q	 GISVWGTT	DANTWLDG
H	NVHGHALV <b>W</b> HS	 AILYYND <b>Y</b> NID-Q	 GISVWGTT	DANTWLT-
Ι	GVRGHTLV <b>W</b> HN	 ALLFYND <b>Y</b> NECFP	 NVTFWGIA	DDHT <b>W</b> LDH
J	KLRGHTFV <b>W</b> HN	 AKLFYND <b>Y</b> NNEMP	 	DRHT <b>W</b> KDN
K	AIRGHNLL-HQ	 AQLFYND <b>Y</b> STENP	 SVSFWGLK	DDYS <b>W</b> LQ-
L	SLRGHTLVWHQ	 AKLFYNDYSTEDP		DDYSWLR-
М	GIRGHTLVWHN	 AKLFYND <b>Y</b> NTEIS	 SVTFWGLK	DDYSWLR-
N	KVRGHTLV <b>W</b> HN	 AKLFYND <b>Y</b> NLEDP	 GVTF <b>W</b> GVA	.DDYT <b>W</b> LYF
0	VIRGHTLVWHN	 AKLFYND <b>Y</b> NTFEP	 NVTF <b>W</b> GLK	DDYSWRAT
Ρ	VVRGHTLV <b>W</b> HN	 AKLFYND <b>Y</b> NTFEP	 	DDYS <b>W</b> RAT
Q	DIRFHTLVWHS	 IKLYMND <b>Y</b> NTEVE		DNHTWLDS
R	ELRFHTLVWHS	 AKLYIND <b>Y</b> NTEVP	 SVTFWGIA	.DNHT <b>W</b> LDG
S	ELRFHTLVWHN	 IKLYIND <b>Y</b> NTDDP	 AVVFWGIS	DKYSWLNG
т	KMRGHTLL <b>W</b> HN	 MKLFINDYNIENN	 AVVFWGVS	DDVTWLS-
U	KMRGHTLLWHN	 MKLFINDYNIENN	 	DDVTWLS-
V	AIIGHCLI <b>W</b> HS	 AELYYNDYSMAQP		.DQNS <b>W</b> RND
W	TVHGHCLVWHS	 AELYYNDYSMSIP	 RVTV <b>W</b> GVE	DGSSWLNG
Х	CFNFHVLL <b>W</b> GA	 TRLMINDYGILSS	 	
Y	CFNFHVLL <b>W</b> GA	 TRLMINDYGILSS	 PAMVWLRE	FMEAYRES
Z	ELYGHTLV <b>W</b> HS	 AKLCIND <b>Y</b> NVEGI	 	DKYS <b>W</b> VP-
AA	IVHGHTLV <b>W</b> HN	 AILIYND <b>Y</b> NIEEI	 AIQFWGFT	DKYS <b>W</b> VPG
AB	QMRGHTLIWHN	 ALLFYNDYNIEDL	 -FVMWGFI	DKYTWIPG
AC	TVHGHALV <b>W</b> HP	 AELYYNDFNTEEN	 GITVWGIA	DPDSWLYT
AD	LLRGHNIC <b>W</b> DS	 MKLCINDYNIETV	 GVSIWQFA	DPTSWIPG
AE	KMRGHVLV <b>W</b> HN	 VALFYNDYNVFLP	 CVTVFGIC	DDYPLYKN
AF	AMRGHTLV <b>W</b> HN	 VSLFYND <b>Y</b> ETAQP	 SVTFWNLI	DENSWLSG
AG	AVRGHTLVWHS	 CKLYYNDYNEYWD	 AVCVWGPN	IDANT <b>W</b> LGS
AH	KMRGHTLV <b>W</b> HS	 VKLFYND <b>Y</b> NTY	 CITWWGPS	DAETWIRN
AI	RARGHCVFWST	 AALFVND <b>Y</b> NVECG	 -IVF <b>W</b> GIM	IQGKM <b>W</b> RKD

Fig. 1. Amino acid sequence alignment of family 10 xylanases. Only the corresponding regions of the equivalents of W85, Y172, W266 and W274 of S.lividans XlnA are shown. The four highly conserved aromatic residues are highlighted in bold. Sequences are identified as follows with the NCBI accession numbers in brackets: A, Streptomyces lividans XlnA [322292]; B, Penicillum chrysogenum XylP [731178]; C, Aspergillus niger XynA [83659]; D, Fusarium oxysporum GunF [1170139]; E, Actinomadura sp.FC7 Xyl1 [478982]; F, Streptomyces thermoviolaceus Xyn1 [2055278]; G, Pseudomonas fluorescens cellulosa XynF [2120656]; H, Cellvibrio mixtus XynB [2144184]; I, Bacillus stearothermophilus XynA [2126856]; J, Caldocellum saccharolyticum XynA [279947]; K, Anaerocellum thermophilum Xyn [1208895]; L, Caldicellulosiruptor XynA [311189]; M, Caldocellum saccharolyticum CelB[121812]; N, Dictyoglomus thermophilum XynA [2494332]; O, Thermotoga maritima XynA [2120364]; P, Thermotoga neapolitana XynA [603892]; Q, Bacillus stearothermophilus Xyn [2126857]; R, Bacillus sp. C125 XynA [80173]; S, Clostridium stercorarium XynB [729101]; T, Thermoanaerobacterium saccharolyticum XynA [538957]; U, Clostridium thermocellum XynX [586269]; V, Bacteroides ovatus XynA [1364168]; W, Prevotella ruminicola XynA [806576]; X, Rhodothermus marinus Xyn2 [1332581]; Y, Rhodothermus marinus Xyn1 [1877399]; Z, Cellulomonas fimi Cex [121856]; AA, Thermotoga neapolitana XynB [2494335]; AB, Clostridium thermocellum XynZ [80546]; AC, Pseudomonas fluorescens cellulosa XynA [77715]; AD, Filobasidium floriforme Xyn [83855]; AE, Caldocellum saccharolyticum ORF4 [139864]; AF, Butyrivibrio fibrisolvens XynB [97313]; AG, Clostridium thermocellum XynY [1084235]; AH, Butyrivibrio fibrisolvens XynA [97316]; AI, Horde vulgare XynII [1718238].

genesis was used to replace these aromatic residues in XlnA and the purified mutant proteins were characterized.

# Materials and methods

## Site-directed mutagenesis

Site-directed mutagenesis of *xln*A was performed according to the method of Kunkel (1985) on phagemid pIAF217 or pAM19.1 (Moreau *et al.*, 1994). The following oligonucleotides were used for mutagenesis: W85A, 5'-CTG GGA GTG <u>GAA</u> GGC GGC CAG GGT-3'; W85H, 5'-CTG CTG GGA GTG <u>GTA</u> GGC CAG GGT-3'; W85H, 5'-CTG CTG GGA GTG <u>GTG</u> GGC CAG GGT GTG-3'; Y172A, 5'-CTC GAC GTT <u>GGC</u> GTC GTT GTA-3'; Y172F, 5'-CTC GAC GTT <u>GAA</u> GTC GTT GTA-3'; Y172S, 5'-CTC GAC GTT <u>GAA</u> GTC GTT GTA-3'; W266A, 5'-GCG CAC ACC <u>GGC</u> GAC GGT GAT-3'; W266F, 5'-GCG CAC ACC <u>GAA</u> GAC GGT GAT-3'; W266H, 5'-GCG CAC ACC <u>GTG</u> GAC GGT GAT-3'; W274A, 5'-CTC CGA CCG <u>GGC</u> GGA GTC GCT-3'; W274F, 5'-CTC CGA CCG <u>GAA</u> GGA GTC GCT-3'; W274H 5'-CTC CGA CCG <u>GTG</u> GGA GTC GCT-3' [underlining indicates the substituted nucleotide(s)]. Screening and subcloning of the mutated genes from pIAF217 or pAM19.1 into plasmids pIAF18 and pIAF18.1, respectively, were done as described previously (Moreau *et al.*, 1994; Roberge *et al.*, 1997a).

## Enzyme production and purification

XlnA was obtained as described previously (Bertrand et al., 1989). Proteins in the supernatant of S.lividans culture were first concentrated by ultrafiltration with a 3 kDa cut-off membrane (Omega). The concentrated proteins were then precipitated with ammonium sulfate at 65% saturation and, after centrifugation, the precipitate was dissolved in 50 mM sodium citrate buffer, pH 6.0. Samples of 100 mg of protein were loaded on a Phenyl-Sepharose column (Pharmacia) in 50 mM sodium citrate buffer, pH 5.6, containing 1 M ammonium sulfate. Proteins were eluted with a decreasing linear gradient to 0 M ammonium sulfate, followed by an increasing linear gradient to 50% ethylene glycol. Protein concentration was measured by UV absorption at 280 nm. The fractions containing XlnA were pooled, dialysed against Milli-Q water and freeze-dried. Further purification to apparent homogeneity as indicated by SDS-PAGE analysis (>95% by Coomassie Brilliant Blue staining), was achieved by separation on a Superdex HR75 beaded column ( $3 \times 60$  cm) (Pharmacia) with 100 mM sodium citrate, pH 6.0, as the eluent. The purified XlnA-containing fractions were pooled, dialyzed and freezedried.

# Enzymatic activity determinations

The specific activity was determined by incubating the enzymes with 4.5 mg/ml birchwood xylan (Sigma) in 50 mM sodium citrate buffer, pH 6.0, at 60°C for 10 min. The released reducing sugars were determined by the *p*-hydroxybenzoic acid hydrazide method (Lever *et al.*, 1984) adapted for microtiter plates. For the determination of Michaelis–Menten constants, the initial velocities of the enzymes were measured at 60°C in 50 mM sodium citrate buffer, pH 6.0, with birchwood xylan concentrations increasing from 0.045 to 4.5 mg/ml. The kinetic parameters were calculated with GraFit software version 3.09b. All enzymatic activities were expressed in international units (IU), where 1 IU represents the amount of enzyme releasing 1  $\mu$ mol/min of reducing sugars using xylose as standard.

# Enzyme stability

For the determination of the stability, solutions of 500  $\mu$ g/ml of enzyme were incubated at 60°C. Samples were taken at regular intervals and kept on ice for 30 min prior to determination of residual activity. The half-life was obtained by plotting the natural logarithm of the residual activity as a function of incubation time. Stability was also determined in the presence of substrate by incubating the enzyme with 7.2 mg/ml birchwood xylan in 50 mM sodium citrate buffer, pH 6.0. At regular intervals, aliquots were withdrawn and analyzed for their reducing sugar contents. The XlnA half-lives were determined by non-linear analysis of the released reducing sugars versus time plots.

# Circular dichroism (CD)

Solutions of 500  $\mu$ g/ml xylanase in 10 mM sodium phosphate buffer, pH 6.0, were analyzed, at room temperature, using a

0.05 cm jacketed cell on a Jasco J-710 spectropolarimeter interfaced with an IBM computer. Data were averaged from 10 acquisitions between 250 and 190 nm at a scan rate of 100 nm/min. Thermal denaturation of XlnA was done by heating 500  $\mu$ g/ml samples in a 0.05 cm jacketed cell at a rate of 0.5°C/min from 50 to 85°C using a Neslab 110 water-bath. Changes in XlnA structure due to unfolding were followed at 210 nm at every 0.2°C. The values were then transformed into fraction of unfolded protein according to the following equation, assuming a two-state unfolding mechanism:

$$f_{\rm D} = (X - X_{\rm n})/(X_{\rm d} - X_{\rm n})$$

where  $X_n$  is the value for folded protein,  $X_d$  the value for unfolded protein and X the value observed (Pace *et al.*, 1989).

#### HPLC analysis of hydrolysis products

Wild-type XlnA (14.8 nM) or the mutant proteins (equivalent amount of enzymatic units) were incubated with birchwood xylan (4.5 mg/ml) or xylopentaose (4.5 mM) in 50 mM sodium citrate buffer, pH 6.0, at 60 or 45°C, respectively. Because of their low thermal stability, the hydrolysis of xylan by mutants W266A and W266F was performed at 45°C along with the wild-type enzyme for comparison purposes. At time intervals, aliquots were withdrawn, boiled for 5 min and diluted 5or 41-fold in water for xylan or xylopentaose hydrolysis, respectively, prior to HPLC analysis. Samples of 100 µl were injected on to a Dionex CarboPac PA1 (4×250 mm) anionexchange column installed on a Dionex DX-500 HPLC system equipped with a pulsed-electrochemical detector interfaced to an IBM computer and a Thermo Separation Products AS3500 autosampler with 150 mM NaOH (1 ml/min) as the eluent. The oligosaccharides were separated by applying a linear gradient to 500 mM sodium acetate in 150 mM NaOH. The identification and concentration determination of the oligoxylosides produced were achieved by comparison with different concentrations of purified standards  $(X_1 \text{ to } X_9)$ analyzed under the same conditions.

#### Results

To study the role of W85, Y172, W266 and W274 in XlnA, three mutant proteins were generated at each position: W85A/ F/H, Y172A/F/S, W266A/F/H and W274A/F/H. Far-UV CD was first used to determine whether any of these mutations would affect the structure of the enzyme. None of the mutations significantly modified the CD spectrum of the enzyme, indicating that the secondary structure content remained the same after mutation (data not shown). Therefore, loss of function and differences with the wild-type protein characteristics described thereafter may only be due to minor local structural changes undetectable by far-UV CD.

To evaluate the consequences of the mutations on the enzyme, the specific activity against xylan was measured. Only mutation Y172F did not alter the activity and all the others decreased it by more than 74% (Table I). Particularly, the three mutations at position W266 (W266A/F/H) dramatically reduced the specific activity by 99.7, 85.7 and 97.5%, respectively, suggesting that this residue is important for the catalytic properties of the protein (Table I). Most of these mutant proteins retained enough activity for accurate evaluation of their kinetic parameters. The catalytic constant ( $k_{cat}$ ) of the enzyme was reduced by values similar to the specific activity (Table I). Under the conditions used, the specific activity is a measure of the apparent  $k_{cat}$  since a saturating concentration

<b>Table I.</b> Specific activity and kinetic parameters of the mutant xylanases
determined with birchwood xylan as substrate

Enzyme	Specific activity (IU/mg)	$k_{cat}$ (s <sup>-1</sup> )	K <sub>M</sub> (mg/ml)	$k_{\text{cat}}/K_{\text{M}}$ (s <sup>-1</sup> .ml/mg)	$\Delta (\Delta G)^{a}$ (kJ/mol)
Wild-type	174	134	0.10	1327	_
W85A	29.4	32.3	0.16	202	5.2
W85F	43.9	50.1	0.10	511	2.6
W85H	31.1	34.1	0.30	114	6.8
Y172A	45.2	16.7	0.51	41.8	9.6
Y172F	180	124	0.24	517	2.6
Y172S	24.8	20.2	0.80	25.2	11.0
W266A	0.6	nd <sup>b</sup>	nd	nd	nd
W266F	24.9	46.0	2.12	21.7	11.4
W266H	4.4	nd	nd	nd	nd
W274A	10.2	8.3	0.91	9.12	13.8
W274F	21.6	22.4	0.18	124	6.6
W274H	21.2	21.2	0.54	39.3	9.8

 $<sup>^{</sup>a}\Delta(\Delta G) = -RT[(k_{cat}/K_{M})_{mutant}/(k_{cat}/K_{M})_{wild-type}]$ 

<sup>b</sup>nd, parameter not determined.

of substrate was used. However, the  $k_{cat}$  value provides a more accurate measure of the catalytic efficiency of XlnA, because enzyme stability cannot influence this parameter. As expected, the binding affinity of XlnA for xylan was reduced by most of the mutations, confirming that these residues play a role in substrate binding. Only mutation W85F did not alter this parameter. Interestingly, replacement of the three tryptophan and the tyrosine residues by a phenylalanine is less detrimental to the affinity of XlnA for xylan (lower  $K_M$  values) than other mutations (Table I), suggesting that the aromatic nature of these residues is important. The results show that mutation of these aromatic residues caused a decrease of 61.0–99.3% in the specificity of XlnA for xylan, which is reflected by losses in binding energy [ $\Delta(\Delta G)$ ] (Wilkinson *et al.*, 1983) ranging from 2.61 to 13.8 kJ/mol (Table I).

The role of these aromatic residues for the thermal stability of XlnA was investigated by activity measurements and CD. The transition temperatures  $(T_m)$  of wild-type and mutant XlnA were determined from the thermal unfolding curves obtained from CD measurements at 210 nm, the wavelength where the difference in ellipticity between the native and unfolded states of XlnA is maximal. Under these conditions, this experiment provides a way to quantify the effects of the point mutations on the stability of the secondary structures present in XlnA. The three mutations introduced at each position have different effects on the thermal stability of XlnA. Mutations W85H, Y172F and W274F/H did not significantly modify the  $T_{\rm m}$  of XlnA, while mutations W85A, Y172S and W274A increased it by 2.0, 1.6 and 4.1°C, respectively. All the other mutations decreased the  $T_{\rm m}$  by values ranging from 3.3 to 7.3°C (Table II). These variations in  $T_{\rm m}$  translate into differences in free energy of unfolding ranging from a stabilization of 9.5 kJ/mol to a destabilization of 16.7 kJ/mol. However, if one generalizes these results, mutations at positions W85, Y172 and particularly W266 tend to decrease the thermal stability of XlnA, whereas mutations at position W274 have no effect or slightly increase the  $T_{\rm m}$  of the protein (Table II).

The thermal stability of mutated proteins was also evaluated by measuring the half-life of the xylanolytic activity at 60°C in absence or in presence of xylan. Since the activity of the enzyme is measured, these assays provide a way to evaluate the effect of the mutation on the thermal stability of the active site loops of XlnA. Therefore, these results yield additional

Table II. Stability parameters of the mutant xylanases							
Enzyme	$T_{\rm m}$ (°C)	$\Delta(\Delta G)^{a}$ (kJ/mol)	Half-life at 60°C (min)				
			Without substrate	With substrate			
Wild-type	68.0	_	131	318			
W85A	70.0	4.5	39	278			
W85F	64.0	-9.2	4	229			
W85H	67.0	-2.2	22	216			
Y172A	63.0	-11.6	4	5			
Y172F	67.1	-2.1	21	238			
Y172S	69.6	3.6	5	42			
W266A	63.3	-10.8	12	166			
W266F	64.7	-7.6	2	2			
W266H	60.7	-16.7	5	122			
W274A	72.1	9.5	57	159			
W274F	68.8	1.8	27	357			
W274H	68.2	0.6	27	236			

<sup>a</sup> $\Delta(\Delta G) = \Delta H_{\text{Tm wt}}(T_{\text{m wt}} - T_{\text{m mut}})$  (Pace *et al.*, 1989).

information on the role of these amino acids located in the active site of XlnA. In absence of substrate, the half-life of XlnA at 60°C was decreased by all the mutations. Replacement of W266 had the worst effect, reducing the half-life by 91-98.5%, whereas mutations at position W274 had a less pronounced effect than the others decreasing the half-life of the enzyme by 56-79% (Table II). In presence of xylan, no protection against thermal inactivation of XlnA was observed for mutations Y172A and W266F, while the other enzymes were protected to levels ranging from 13 to 112% of the wildtype value (Table II). The degree of protection against thermal inactivation provided by the presence of xylan, in comparison with the wild-type protein, is dependent on the enzyme affinity for xylan, i.e. the half-life in the presence of substrate is proportional to the affinity for xylan, as suggested previously (Roberge et al., 1998). This, however, cannot explain the subtle variations observed for some mutations such as Y172A. In that particular case, one would expect Y172A to be more stable than Y172S since its  $K_{\rm M}$  is lower, but this is not the case (Table II). This deviation from the general rule could be explained by the important decrease in stability of the structure of that mutant as measured by CD, probably caused by local structural perturbations (Table II). Nevertheless, these and previous results (Roberge et al., 1998) support the fact that stabilization provided by the presence of xylan is dependent on the affinity of the enzyme for its substrate.

In order to verify whether the mode of action of XlnA was affected by any of the mutations studied, the hydrolysis products of birchwood xylan and X<sub>5</sub> were analyzed by HPLC. Figure 2 shows the results obtained with wild-type XlnA. In our HPLC analysis, oligoxylosides with a degree of polymerization <10 were well resolved and linearly produced throughout the experiment. The major oligoxylosides produced in the hydrolysis of birchwood xylan by XlnA are X<sub>3</sub> and X<sub>4</sub> (Figure 2A). No xylose is seen after 30 min of hydrolysis, while XlnA also produced many other oligoxylosides such as X2, X5-X8 and other oligoxylosides with degrees of polymerization up to 20 (Figure 2A). From the HPLC analysis at different intervals of incubation, a time course of oligoxylosides production was constructed and the results for wild-type XlnA are shown in Figure 2C. Oligoxylosides with a degree of polymerization <9 were analyzed and their rates of production decreased in the order  $X_3 > X_4 > X_5 > X_2 = X_6 > X_7 > X_8$ . Using the same method, hydrolysis of xylan by the different mutant proteins was also analyzed. No major differences from the wild-type profile were observed (data not shown). However,  $X_2$  was produced by all the mutant proteins at a rate 20–60% slower than that of the wild-type XlnA.

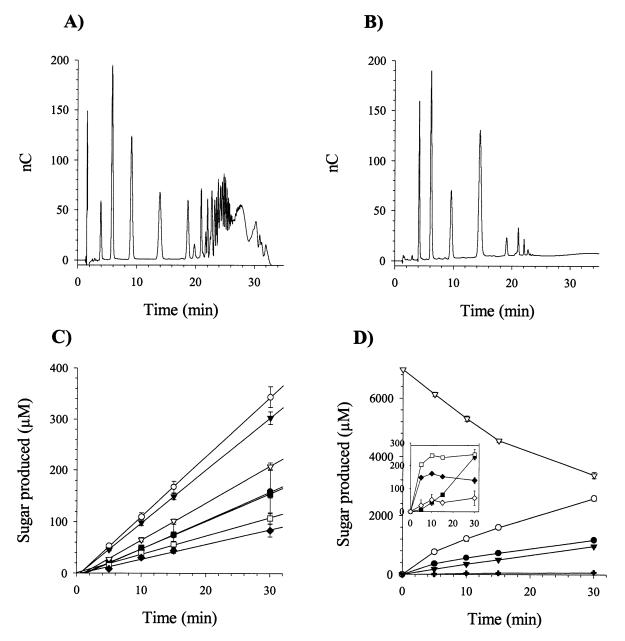
Hydrolysis of X<sub>5</sub> mainly produces X<sub>3</sub> but also significant amounts of X<sub>2</sub> and X<sub>4</sub> and only very little X<sub>1</sub> is observed after 30 min of hydrolysis (Figure 2B). Incubation of XlnA with a saturating amount of X<sub>5</sub> also gives rise to transglycosylation products with degrees of polymerization up to 9 (Figure 2D, inset). In the hydrolysis of  $X_5$ , the increase in the concentrations of released oligoxylosides was not linear over time during the 30 min period used in our experiment (Figure 2D) and it was impossible to calculate the rate of production for each oligoxyloside as in the case with the hydrolysis of xylan. The amount of the transglycosylation products  $X_7-X_9$  increased in the first 5 min and was stable for the remainder of the 30 min period, while the concentration of X<sub>6</sub> increased with time during the whole period (Figure 2D). Since  $X_1$  is not readily produced from the hydrolysis of X<sub>5</sub>, the transglycosylation product  $X_6$  was not observed at first, but was probably generated by the assembly of two  $X_3$  or one  $X_2$  and one  $X_4$ in the course of the reaction. Neither the hydrolysis nor the transglycosylation products of the reaction with X<sub>5</sub> were significantly altered by any of the mutations studied, indicating that the mode of action of XlnA was not affected (data not shown). However, when X<sub>5</sub> was hydrolyzed with the same quantity of enzymatic units, mutations Y172A, Y172S, W266F W274A and W274H caused a reduction in hydrolysis efficiency by 88, 72, 91, 73 and 51%, respectively (data not shown).

## Discussion

S.lividans XlnA is a well characterized member of family 10 glycosyl hydrolases for which the three-dimensional structure of the catalytic domain is known (Derewenda et al., 1994). There are over 30 enzymes in this family, four of which have known three-dimensional structures (Henrissat and Bairoch, 1998). Thus, primary and tertiary structure comparisons are very useful tools in the identification of potentially important residues. These comparisons have shown the presence of conserved aromatic residues in the active site of this family that could be important in substrate binding. Site-directed mutagenesis was used to replace specifically four of these aromatic amino acids shown to be located in subsites -1 and +1 of the active site of family 10 xylanases (Harris et al., 1994). While mutation of either of these four active residues did not alter the hydrolysis pattern of the enzyme with either xylan or xylopentaose as substrates, significant variations in kinetic and stability parameters from the wild-type were observed, allowing a better understanding of the roles of these four aromatic residues in the structure-function relationships of family 10 xylanases.

## Role of tryptophan 85

In the structure of XlnA, W85 (Nɛ1) is hydrogen-bonded to E128 (Oɛ1), the acid–base catalyst and to N127 (O $\delta$ 1) (Figure 3). This pattern was also observed in the structures of other family 10 xylanases, which suggested its importance for catalysis by the enzyme (Dominguez *et al.*, 1995; Harris *et al.*, 1996; White *et al.*, 1996). From the kinetic results, the 70% average loss in activity due to the three mutations tested (Table I) could be attributed to secondary effects involving one or both of these residues, which have been shown to be very



**Fig. 2.** HPLC analysis and time course of the products obtained from the hydrolysis of, respectively, birchwood xylan (A and C) and xylopentaose (B and D) by wild-type XlnA. The oligoxylosides are identified as follows with their retention times in parentheses:  $X_1$ , + (2.9 min);  $X_2$ ,  $\bigcirc$  (4.2 min);  $X_3$ ,  $\spadesuit$  (6.0 min);  $X_4$ ,  $\blacktriangledown$  (9.3 min);  $X_5$ ,  $\bigtriangledown$  (14.2 min);  $X_6$ ,  $\blacksquare$  (18.9 min);  $X_7$ ,  $\square$  (21.1 min);  $X_8$ ,  $\blacklozenge$  (22.2 min);  $X_9$ ,  $\diamondsuit$  (22.9 min).

important in the catalytic efficiency of XlnA (Moreau *et al.*, 1994; Roberge *et al.*, 1997b). Moreover, the aromatic nature of W85 appears to play a role in substrate binding, as expected. Increases in  $K_{\rm M}$  were obtained by replacing W85 by non-aromatic residues (A and H), while the mutant enzyme W85F maintained its affinity for xylan (Table I). This suggests that W85 is involved in stacking interactions with the substrate. Furthermore, this residue is important for the thermal stability of the protein, as shown by the decrease in mutant enzyme half-lives at 60°C in the absence of substrate (Table II). In that case, the aromatic nature of W85 is probably not the only factor involved since the phenylalanine substitution seems to have the worst effect on that parameter. These results suggest that hydrogen bonding involving W85 is important for the stability of the protein.

## Role of tyrosine 172

The hydroxyl group of Y172 is involved in hydrogen-bonding interactions with D132 (O $\delta$ 1), R139 (N $\eta$ 1) and an active-site water molecule (Figure 3). However, these interactions do not seem to be important for enzyme activity since Y172F mutant protein, without hydrogen-bonding possibility, is as active as the wild-type protein (Table I). The aromatic nature of this residue is important in substrate binding, since its replacement by non-aromatic residues decreases the affinity for xylan (Table II) and the apparent rate of hydrolysis of xylopentaose, while the Y172F mutation behaves similarly to the wild-type. Our results also show that Y172 is important for enzyme stability, particularly in the absence of substrate at 60°C (Table II). Again, the Y172F replacement is less detrimental to enzyme stability in the presence of substrate, strengthening the hypo-

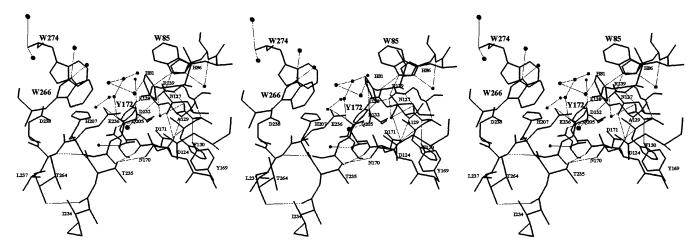


Fig. 3. Environment of the two catalytic residues in *S.lividans* XlnA showing the positions of the aromatic amino acids. Hydrogen bonds are represented by dashed lines and water molecules by filled spheres.

thesis that the protection gained against thermal inactivation is highly dependent on substrate affinity (Roberge *et al.*, 1998). However, results for the  $T_{\rm m}$  of the proteins suggest that hydrogen-bonding interactions are not important for the structural stability of XlnA, since similar  $T_{\rm m}$  values were obtained for both the Y172F mutant and wild-type proteins (Table II). In this case, it is difficult to draw conclusions on the role of the aromatic nature of Y172 without knowing the structure of the mutant protein, since the Y172S mutation did not decrease the  $T_{\rm m}$  of XlnA while significantly reducing the half-lives with and without substrate (Table II).

## Role of tryptophan 266

Amino acid W266 is also highly conserved in family 10 xylanases. Its Nɛ1 atom is hydrogen-bonded to two active-site water molecules. However, W266 is also stacked to E236, the residue acting as a nucleophile in the hydrolysis of substrate in XlnA (Figure 3). This interaction was proposed to be of importance in other families of glycosyl hydrolases using the same mechanism of action (Dominguez *et al.*, 1995). Our results confirm that these structural features are indeed important for the function of this type of enzyme and comparison of the specific activity of the different mutant proteins at this position suggests that the aromatic nature of W266 is involved in the activity of the protein.

The equivalents of residue W266 were shown to be involved in substrate binding in two other family 10 xylanase structures when complexed with substrates (Harris et al., 1996; White et al., 1996). Our kinetic analysis of W266F mutant protein clearly shows that this amino acid is involved in substrate binding and supports the previously published structural analysis. While mutant W266F retains a certain activity level, it is still 66% less active than the wild-type protein (Table I), indicating that W266 probably serves to position the nucleophile in the active site for efficient binding and catalysis. The large variation in affinity for xylan and the important reduction in activity towards xylopentaose exhibited by mutant W266F probably reflect a bad positioning of E236. This would weaken one of the major stabilizing interactions between OE2 of E236 and OH-2 of the substrate at the transition state (White et al., 1996). Moreover, the modest effect on the  $k_{cat}$  of the enzyme caused by the W266F mutation suggests that the hydrogen bond observed between Nɛ1 of the W266 equivalent and the catalytic intermediate analog in C.fimi Cex (White et al., 1996; Notenboom *et al.*, 1998), which is missing in W266F, plays a minor role in the stabilization of the intermediate. Finally, W266 is very important in the thermal stability of XlnA, as shown by the large decreases in half-lives at 60°C and in  $T_m$  exhibited by all the mutant proteins studied (Table II). Again this supports the concept that the residues forming the complex hydrogen bonding network in the active site of the enzyme are involved in interactions crucial for the integrity of the structure of family 10 xylanases.

# Role of tryptophan 274

In contrast to the three other amino acids studied here, W274 is not involved in hydrogen bonding interactions with other residues (Figure 3). In fact, W274 (N) is only hydrogen bonded to two active-site water molecules. Our kinetic analysis of the three mutant proteins at position W274 showed that this residue is involved in substrate binding (Table I). This supports the observations that the equivalent of W274 in the binding of xylopentaose at subsite -1 in XynA from P. fluorescens subsp. cellulosa has a high B value in the refined structure of this enzyme (Harris et al., 1994). The authors suggested that this residue could be important for the binding of large xylan substrates (Harris et al., 1994, 1996). Our results support this hypothesis, although the significant decrease in activity towards xylopentaose caused by mutations W274A and W274H, but not by W274F, indicates that the aromatic nature of this residue is also important for efficient catalysis of smaller oligoxylosides. Comparing the hydrolysis patterns of both xylan and xylopentaose by the three mutants showed that the aromatic nature of this residue plays a large role in substrate binding, suggesting that stacking interactions between W274 and the substrate are involved. From the decreases in  $k_{cat}$ obtained (Table I), W274 also seems to play a role in the stabilization of the catalytic intermediate. This result was expected after analysis of the complexed structures of other family 10 xylanases that suggested W274 to be important for the specificity between glucosyl and xylosyl substrates (White et al., 1996; Notenboom et al., 1998). Finally, our results on the  $T_{\rm m}$  values have shown that W274 is not crucial to the stability of the XlnA structure. Still, replacement of this active site residue reduced the half-life in absence of substrate by up to 79% (Table II), suggesting that the active site environment is affected by these mutations.

In conclusion, this study on four conserved aromatic residues

in family 10 xylanases using site-directed mutagenesis has demonstrated their role in substrate binding and catalysis by this type of enzyme. As expected for sugar-binding proteins, the aromatic nature of W85, Y172 and W274 is particularly important in substrate binding. W266 is also involved in the catalytic efficiency of the protein and the four residues were shown to be important for the thermal stability of XlnA. This kinetic evidence supports the structural observations that were made on other family 10 xylanases and thus improves our knowledge of the structure–function relationship of this type of protein.

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